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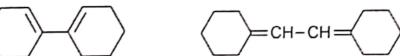
Third Semester M.Sc. Degree Examination, February 2024 Chemistry / Analytical Chemistry / Polymer Chemistry CH 232 / CL 232 / PC 232 : ORGANIC CHEMISTRY – III (2020 Admission Onwards)

Time: 3 Hours Max. Marks: 75

SECTION - A

Answer two among (a), (b) and (c) from each. Each sub question carries 2 marks.

 (a) Among the following dienes which one will observe at lower wavelength? Why?



- (b) An intense peak at m/e: 149 is observed in the mass spectrum of diethyl phthalate. Account for this.
- (c) Two γ lactones having double bond give absorption at 1800 cm⁻¹ and 1750 cm⁻¹. Write the structure and assign the value for each compound.
- 2. (a) An organic compound molecular formula C_4H_9NO shows 1H NMR peaks at δ (ppm): 2.9 (t, 4H), 3.8 (t, 4H), 1.8 (broad s, 1H) assign a suitable structure.
 - (b) The 13 C-NMR spectrum of one of the butyl acetate isomers (C₄H₉OCOCH₃) shows signal at δc 22, 28, 80 and 170 what is its structure? (Intensity of peak at δc 28 much more intense than that δc 22)
 - (c) Predict the chemical shift position for the protons in 4-nitroanIsole.

P.T.O.

3. (a) Complete the following:

(i) Ph CH = CHCoPh + CN
$$\theta$$
 weak base ?

(ii) CH₃ CH = CH - COCH₃
$$\xrightarrow{\text{CH}_3 \text{ MgI}}$$
 $\xrightarrow{\text{H}_2\text{O}, H^+, \text{Strongbase}}$

(b) Predict the structure of product A and B in the reaction given below.

$$\begin{array}{c}
O \\
\hline
 & (i-C_3H_7)_2N \xrightarrow{\text{OLi+}} A \xrightarrow{\text{CH}_2=\text{CH-CH}_2\text{Br}} \\
\hline
 & \text{THF, -78°C} & A \xrightarrow{\text{HMPT, -78°C}} B
\end{array}$$

(c) Predict the product in the following:

$$R-C = O + C_{P2}Ti \xrightarrow{CH_2} AI Me_2 \xrightarrow{} ?$$

$$C_{P2} = Cyclopentadiene$$

- 4. (a) Give any two protecting groups for phenols.
 - (b) Suggest a retro synthetic route for the following compounds.

- (c) Why cyanide ions (CN^-) is a highly specific catalyst for the benzoin condensation?
- 5. (a) What is Jones oxidation reaction?
 - (b) Illustrate the use of borohydride for the transformation of isopropanol to n-propanol.
 - (c) Write the main product of reaction between o-dichlorobenzene and sodamide.

$$(10 \times 2 = 20 \text{ Marks})$$

SECTION - B

Answer either (a) or (b) from each question. Each sub question carries 5 marks.

- (a) Write all Possible isomers of molecular formula C₆H₆O whole UV band exhibit at λ max 187 nm (high intensity) and at λ max 280 nm with low intensity.
 - (b) Show the mass fragmentation pattern of 3-Methyl pyridines.
- 7. (a) Assign the structure of the compound on the basis of spectral data. UV λ max: No band above 200 nm; IR: v max:1740 cm⁻¹ 1 H NMR δ = 4.0 (t, 2, H); 2.2 (m, 2H); 2.3(m, 1H):1.15 (s, 3H)
 - (b) Write a note on shift reagent in NMR spectroscopy.
- 8. (a) What is Gillman reagent? How it's formed? Complete the following.

COCI
$$+ (CH_3)_2 Culi \longrightarrow ?$$

$$NO_2$$

- (b) Write a note on 1,2 and 1, 4- enone addition reactions.
- 9. (a) Explain stork enamine acylation reaction. Outline its mechanism.
 - (b) What is Umpolung? Explain.
- (a) How is Tri-n-butyl tinhybride ((n − Bu)₃Sn₄) prepared where it is used in the field of organic group transformation; Explain it with suitable examples.
 - (b) Explain Birch reduction. Complete the following with mechanism.

 $(5 \times 5 = 25 \text{ Marks})$

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SECTION - C

Answer any three questions. Each question carries 10 marks.

- 11. (a) Explain the EI in mass spectroscopy.
 - (b) Summarize the general trends of structural variation on the position of carbonyl stretching frequency in IR spectra.
- 12. Three isomeric compound, $C_4H_8O_3$ give the following 1H spectra. Deduce the possible structure for them.

Compound 1 : δ 1.3 (t, 3H, J = 7 Hz); 3.6 (q, 2 Hz, J = 7 Hz); 4.15 (s, 2 H); 12.1 (s, 1H) ppm.

Compound 2: 1.2 (d, 3H, J-7 Hz); 2.3 (d,2 H,J = 7 Hz); 4.15 (1:5:10:10:5:1 sextet, 1H, J= 7 Hz) ppm (Spectrum run in D_2O)

Compound 3:3.5 (s, 3H); 3.8 (s, 3H) 4.08 (s, 2H) ppm

- 13. Write notes on:
 - (a) Silane carbanion and its reaction.
 - (b) Alkynyl Cu (I) reagents.
- 14. Explain the following:
 - (a) Creation of cis and trans double bonds.
 - (b) Stepens-Castro coupling reactions.
- 15. Give a brief account on application of following reagents in organic synthesis.
 - (a) OsO₄
 - (b) DIBAL-H

 $(3 \times 10 = 30 \text{ Marks})$

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